

TWO SAPONINS FROM *ZEXMENIA BUPHTHALMIFLORA*

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Key Word Index—*Zexmenia buptharmiflora*; *Wedelia buptharmiflora*; Heliantheae; Compositae; structure elucidation; saponins; oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside; oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside-28-O- β -D-glucopyranoside.

Abstract—Oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside and its 28-O- β -D-glucopyranosyl ester have been isolated from aerial parts of *Zexmenia buptharmiflora*. The chemotaxonomical significance of the transfer of *Wedelia buptharmiflora* to *Zexmenia buptharmiflora* is discussed.

INTRODUCTION

As a part of our investigations on *Wedelia* species we have previously reported on the isolation and identification of di- and triterpenoids from the aerial parts of *W. glauca* [1] and *W. buptharmiflora* [2]. In the meantime *W. buptharmiflora* Lorentz has been reclassified as *Zexmenia buptharmiflora* (Lorentz) Ariza [3]. Both genera are closely related (family: Compositae, tribe: Heliantheae) although *Wedelia* belongs to the subtribe Ecliptinae and *Zexmenia* to Verbesininae [4].

This paper deals with the structure elucidation of two saponins: oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside (1) and oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside-28-O- β -D-glucopyranoside (2) isolated from *Z. buptharmiflora* and identified by spectral and chemical methods.

RESULTS AND DISCUSSION

Column chromatography of the ethanolic extract of aerial parts of *Z. buptharmiflora* yielded three fractions. Fractions 2 and 3 contained the acidic terpenic glycosides 1 and 2, respectively, which were purified by chromatography of their methyl derivatives (3 and 4, respectively).

Upon acid hydrolysis of 3, methyl oleanolate, rhamnose and an uronic acid were obtained whereas 4 afforded oleanolic acid, rhamnose, glucose and an uronic acid. Alkaline hydrolysis of 4 yielded glucose and a glycoside that on methylation with diazomethane led to 3. This fact showed that 4 was a glucoside ester of 3.

Reduction of 4 provided 5 which was hydrolysed to give erythrodilol, rhamnose and glucose and thus providing evidence for the presence of glucuronic acid in 3 and 4. The ^1H NMR spectrum of the TMSi derivative of 3 showed, among other signals, a singlet at δ 5.11 ascribed to the anomeric proton of one moiety of rhamnose [5], a doublet ($J = 6$ Hz) at δ 4.52 of the anomeric proton of one

glucuronic acid moiety with β -configuration [6], two singlets at 3.80 and 3.64 assigned to the methyl esters of the glucuronic acid and the aglycone, respectively. In the ^1H NMR spectrum of the TMSi derivative of 4 a new doublet ($J = 6$ Hz) appeared at δ 5.60 assigned to the anomeric proton of one β -D-glucose [7] while the singlet at 3.64 was absent. Consequently the anomeric hydroxyl of this β -D-glucose was attached to the carboxyl group of the oleanolic acid.

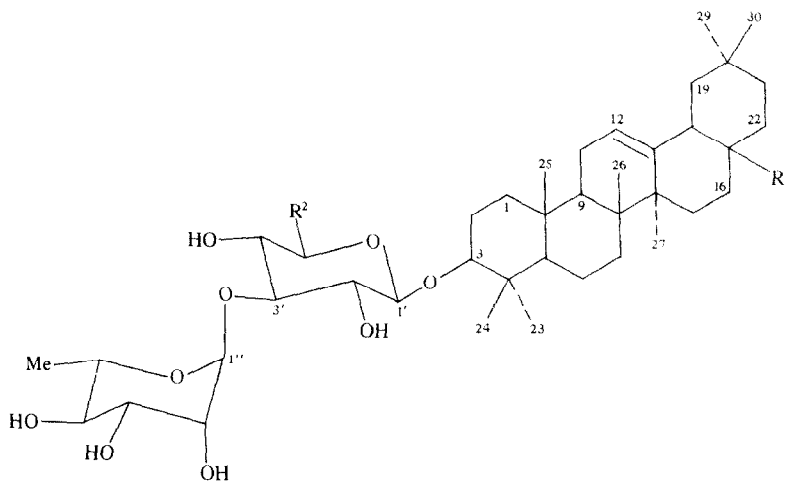
The mass spectrum of the TMSi derivative of 3 showed the presence of a very intense ion at m/z 363 (TMSi rhamnose oxonium ion) indicating that rhamnose was the terminal moiety [8] of the disaccharide attached to the HO-3 of the aglycone.

Evidence of the attachment of the glucose moiety in 4 to the carboxyl group of the oleanolic acid and not to the uronic acid was provided by the ^{13}C NMR spectrum of 4 which showed the presence of a signal for a COOMe at δ 52.1 ascribed to a COOMe of the uronic acid (COOMe of methyl oleanolate: 51.6; COOMe of methyl glucuronate: 52.1). Additionally, the C-28 (C=O) signal of the triterpenoid was affected in 4 due to the glycosylation of this carbon. (Compare C-28 in 3 and 4, Table 1; 178.0 and 176.4, respectively). The latter (176.4) is in agreement with the C-28 chemical shift of other glucosyl esters of oleanolic acid derivatives [9].

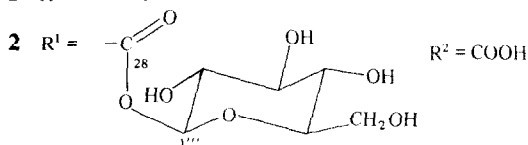
On the contrary, C-6' (COOMe of the methyl glucuronate) in 3 and 4 have the same value, 170.5 and 170.3, respectively. The ^1H NMR spectrum of 3 showed signals at δ 3.80 and 3.64 assigned to COOMe of methyl glucuronate and methyl oleanolate, respectively. The ^1H NMR spectrum of 4 gave a signal at 3.80 due to the methyl glucuronate. Also acid hydrolysis of 4 afforded oleanolic acid. If the glucose moiety was attached to the COOH of the glucuronic acid, methyl oleanolate would have been the hydrolysis product.

The ^{13}C NMR spectra of 3, 4 and 5 (Table 1) showed that all the sugars were in the pyranose form [10, 11] and that rhamnose was α -linked [5] to C-3' of glucuronic acid. Comparison of the ^{13}C NMR spectrum of 5 with that of the 3-O- β -D-glucopyranoside of methyl oleanolate [12, 13] showed deshielding of C-3' (5.3 ppm) and shielding of

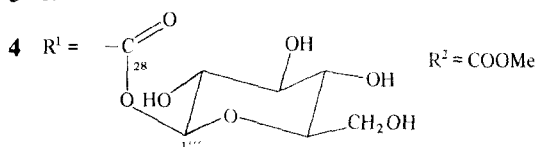
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1 $R^1 = R^2 = \text{COOH}$



3 $R^1 = R^2 = \text{COOMe}$



5 $R^1 = R^2 = \text{CH}_2\text{OH}$

C-4' (2.3 ppm) while C-2' was not affected. Glycosidation of either C-2' or C-4' was not in agreement with the observed upfield shift of C-4'.

Taking into account the acylation effect of secondary hydroxyl groups [14] C-1''' is expected to appear downfield to the signal of the anomeric carbon of β -D-glucopyranose ($\delta 98.9$, $\text{C}_5\text{D}_5\text{N}$). However, in **4** C-1''' is observed at 95.4 in accordance with data reported for other glucopyranosyl esters, e.g. arjunglucosides [9], steviosides and paniculoses [15].

Therefore **1** is oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside and **2** is oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside-28-O- β -D-glucopyranoside. The Klyne rule [16] confirmed the proposed configurations.

The structures were confirmed by chemical methods using total methylation of **3** and **5**. Hakomori methylation [17] and further hydrolysis of **5** afforded 2,3,4-tri-O-methylrhamnose and 2,4,6-tri-O-methylglucose which were identified by comparison with standards. Hakomori

methylation of **3** followed by LiAlH_4 reduction and hydrolysis gave 2,3,4-tri-O-methylrhamnose and 2,4-di-O-methylglucose.

Compounds **1** and **2** have been isolated from seedcoats of *Putranjiva roxburghii* (Euphorbiaceae) [18] but they were only identified by chemical methods. The present paper is the first report on the spectral data (^1H NMR, ^{13}C NMR and MS) of these glycosides and on their occurrence in the family Compositae.

In spite of the similarity of the petrol extracts of *W. glauca* [1] and *Z. buphthalmiflora* [2] the methanolic extracts of both differ considerably since **1** and **2** were not found in *W. glauca* that contains instead atractyloside, which is responsible for its high toxicity to cattle [C. D. Schteingart and A. B. Pomilio, unpublished results].

Sesquiterpene lactones [19–21], sesquiterpenes [21], tricyclic diterpenes [20], kaurenic acid and grandiflorenic acid [20] have been reported in *Zexmenia* species. On the other hand, esters of 15 α -hydroxykaur-16-en-19-oic acid have been frequently found in *Wedelia* species but only

Table 1. ^{13}C NMR spectral data of compounds 3, 4 and 5 (δ , $\text{C}_5\text{D}_5\text{N}$, 20.15 MHz)

Carbon	3	4	5	Carbon	3	4	5
1	39.3	39.1	39.3	1'	106.7	106.5	106.4
2	26.3	25.9	26.4	2'	75.4	75.2	75.8
3	89.3	89.3	89.0	3'	82.0	81.9	83.8
4	38.4	38.5	38.7*	4'	71.2	71.0	69.8
5	55.6	55.6	55.6	5'	76.8	76.6	77.9
6	18.3	18.1	18.5	6'	170.5	170.3	62.6
7	33.0	32.9	32.8				
8	39.5	39.6	40.0*	1''	102.6	102.3	102.8
9	47.7	47.7	47.8	2''	72.1	72.0	72.4
10	36.8	36.6	36.7	3''	72.4	72.0	72.4
				4''	73.8	73.5	73.9
11	23.5	23.4	23.8	5''	69.6	69.5	69.8
12	—	—	—	6''	18.3	18.1	18.2
13	144.0	143.8	145.0				
14	41.8	41.8*	41.9	1'''		95.4	
15	27.9	27.9	26.1	2'''		73.5	
16	23.5	23.4	22.6	3'''		78.2*	
17	46.8	46.7	37.5	4'''		70.7	
18	41.8	41.5*	42.7	5'''		78.6*	
19	46.0	46.7	47.0	6'''		61.8	
20	30.6	30.5	31.1				
				6'-Me	52.1	52.1	
21	33.8	32.9	34.5				
22	33.0	31.9	31.7				
23	27.9	27.9	28.0				
24	16.7	16.6	16.3				
25	15.3	15.3	15.6				
26	17.0	17.2	16.3				
27	26.0	25.9	26.1				
28	178.0	176.4	68.6				
29	33.0	32.9	33.3				
30	23.5	23.4	23.8				
28-Me	51.6						

—, Signal masked by solvent.

* Assignments may be interchanged in each vertical column.

once in *Zexmenia* [2]. Further chemotaxonomic studies on species of both genera would clear up the actual relationship between them.

EXPERIMENTAL

Mps are uncorr. ^1H NMR spectra were recorded at 100 MHz in CDCl_3 , ^{13}C NMR spectra at 20.15 MHz in $\text{C}_5\text{D}_5\text{N}$ and MS at 70 eV by direct inlet.

Plant material. *Zexmenia bupthalmiflora* (Lorentz) Ariza was collected in Padre Búodo (Utracán, Province of La Pampa, Argentina). A voucher specimen Steibel 6105 is deposited in University of La Pampa, Faculty of Agronomy.

Extraction and isolation of 1 and 2. Dried and ground aerial parts (3 kg) were successively extracted with petrol (69 g of extract, 2.4% of dry plant) and EtOH (151 g of extract, 5.2% of dry plant).

A portion of the ethanolic extract of *Z. bupthalmiflora* was chromatographed on a silica gel H (Merck) column using a gradient of MeOH in CHCl_3 as eluent. Three main fractions were obtained. Fraction 1 contained minor terpenoids while fractions 2 and 3 had mainly glycosides 1 and 2, respectively; TLC (silica gel, 1: R_f 0.51, 2: R_f 0.24; CHCl_3 -MeOH-HOAc- H_2O , 30:8:1.5:1; anisaldehyde- H_2SO_4). Fractions 2 and 3 were sep-

arately methylated with CH_2N_2 in Et_2O -MeOH and further purified by silica gel H column chromatography with CHCl_3 -MeOH (30:2) to give the methyl derivative of 1 (3, 30 mg) and with a gradient of CHCl_3 -MeOH (30:3 to 30:7) to yield the methyl derivative of 2 (4, 190 mg); TLC (silica gel, 3: R_f 0.45, 4: R_f 0.15; toluene-EtOH, 20:7; anisaldehyde- H_2SO_4).

Identification of the glycosides. Dimethyl ester of oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside (3). Mp 189–192° (MeOH- H_2O , 1:1); $[\alpha]_D^{20}$ -10°. (MeOH, c 0.20); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3370, 1740, 1630, 1000, 860, 760; ^1H NMR (TMSi deriv.): δ 5.30 (1H, m, H-12), 5.11 (1H, s, H-1''), 4.52 (1H, d, J = 6 Hz, H-1'), 4.1–3.5 (sugar protons), 3.80 (3H, s, 6'-COOMe), 3.64 (3H, s, 28-COOMe), 1.15 (3H, s, Me-27), 0.94 and 0.92 (12H, two br s, Me-23, 25, 29 and 30), 0.82 (3H, s, Me-24), 0.75 (3H, s, Me-26); ^{13}C NMR: see Table 1; MS m/z (rel. int.): 470 $[\text{A}]^+$ (6), 453 $[\text{A}-\text{OH}]^+$ (7), 452 $[\text{A}-\text{H}_2\text{O}]^+$ (9), 410 $[\text{A}-\text{HCOOMe}]^+$ (5), 262 (RDA in ring C) (70), 203 $[\text{262}-\text{COOMe}]^+$ (100), 189 $[\text{203}-14]^+$ (40); MS (TMSi deriv.) m/z (rel. int.): 527 $[\text{A}+\text{TMS}-\text{H}-15]^+$ (2), 470 $[\text{A}]^+$ (0.5), 453 $[\text{A}-\text{OH}]^+$ (43), 452 $[\text{A}-\text{H}_2\text{O}]^+$ (13), 363 (TMSi rhamnose oxonium ion) (74), 273 $[\text{363}-\text{TMSiOH}]^+$ (15), 262 (40), 204 (100), 203 (43), 189 (20); A = aglycone.

Acid hydrolysis of 3. This was carried out in a sealed tube with 7% H_2SO_4 for 3 hr at 100°. The aglycone was extracted with

CHCl_3 and identified as methyl oleanolate by comparison with standards (TLC, mp, mmp, GC). The aq. layer was neutralized with BaCO_3 and successively treated with a strong cationic exchange resin (AG 50W-X8, Bio Rad, form: H^+) and a weak anionic exchange resin (AG 3-X4A, Bio Rad, form: $-\text{NH}_2$). Analysis of the eluate by TLC (Cellulose F, Merck, 100 μm ; n -BuOH- $\text{C}_5\text{H}_5\text{N}$ - H_2O , 6:4:3; AgNO_3 -2% NaOH) with appropriate standards showed the presence of rhamnose. Elution of the anionic resin column with 7% H_2SO_4 , neutralization with BaCO_3 and further treatment with a strong cationic resin (form: H^+) led to a soln that contained glucuronic acid (TLC, cellulose, as above).

Methyl ester of oleanolic acid 3-O-(3'-O- α -L-rhamnopyranosyl)- β -D-glucuronopyranoside-28-O- β -D-glucopyranoside (4). Mp 234–237° (MeOH- H_2O , 1:1); $[\alpha]_{\text{D}}^{20}$ -17° (MeOH, c 0.25); ^1H NMR (TMSi deriv.): δ 5.60 (1H, d, J = 6 Hz, H-1'''), 5.33 (1H, m, H-12), 5.14 (1H, s, H-1''), 4.48 (1H, d, J = 6 Hz, H-1'), 4.1–3.4 (sugar protons), 3.80 (3H, s, 6'-COOMe), 1.12 (3H, s, Me-27), 0.92 (12H, br s, Me-23, 25, 29 and 30), 0.82 (3H, s, Me-24), 0.78 (3H, s, Me-26); ^{13}C NMR: see Table 1; MS m/z (rel. int.): 456 $[\text{A}]^+$ (4), 439 $[\text{A} - \text{OH}]^+$ (4), 438 $[\text{A} - \text{H}_2\text{O}]^+$ (5), 248 (RDA in ring C) (100), 207 (rings A and B) (48), 203 $[248 - \text{COOH}]^+$ (85), 189 (42).

Acid hydrolysis of 4. This was carried out as above yielding oleanolic acid (mp, TLC, GC, MS), rhamnose, glucose and glucuronic acid.

Alkaline hydrolysis of 4. Compound 4 was heated in a sealed tube with NH_3 (c) for 2 hr at 100°. The soln was acidified with aq. H_2SO_4 and twice extracted with n -BuOH. The organic layer was evaporated to dryness, methylated with CH_2N_2 in Et_2O -MeOH and purified by silica gel CC (toluene-EtOH, 20:4) to give 3. The aq. layer was neutralized, worked up and analysed as mentioned for 3. Only glucose was detected.

Preparation of 5. Compound 4 (100 mg) was treated with LiAlH_4 in excess under reflux in THF for 16 hr. The suspension was worked up in the usual way and the product was purified by silica gel H CC (toluene-EtOH, 20:4) yielding 47 mg of 5 (60%); TLC (silica gel, R_f 0.31, toluene-EtOH, 20:7, anisaldehyde- H_2SO_4); $[\alpha]_{\text{D}}^{20}$ +4° (EtOH, c 0.43); ^{13}C NMR: see Table 1.

Acid hydrolysis of 5. As above. Erythrodilol (TLC, GC with standards), glucose and rhamnose (TLC and GC of the acetate alditols with standards) were detected.

Total methylation of 5 and characterization of the partially methylated sugars. Compound 5 was methylated by the Hakomori method. The product was purified by silica gel H CC (toluene-EtOAc, 4:1) and hydrolysed in a sealed tube with Kiliani mixture (HOAc - HCl - H_2O , 35:15:50) for 16 hr at 80°. MeOH was added to the soln which was neutralized with the weak anion resin (form: $-\text{NH}_2$) in MeOH and analysed by TLC (silica gel; toluene-EtOH, 20:7) with partially methylated sugars as standards. 2,3,4-Tri- O -methylrhamnose (R_f 0.74) and 2,4,6-tri- O -methylglucose (R_f 0.35) were found.

Total methylation of 3, reduction and analysis of the partially methylated sugars. Compound 3 was dissolved in DMSO and treated with dimethyl carbanion for 2 hr at room temp. MeI was added and the soln left standing for 1.5 hr at room temp. The

product was isolated in the usual manner, reduced under reflux with LiAlH_4 in THF overnight and hydrolysed as described for 5. Partially methylated sugars were analysed by HPTLC (silica gel, Merck, toluene-EtOH, 20:13) with standards. 2,3,4-tri- O -methylrhamnose (R_f 0.90) and 2,4-di- O -methylglucose (R_f 0.45) were detected. GC identification of the acetate alditols with authentic samples was also performed.

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REFERENCES

- Oberti, J. C., Pomilio, A. B. and Gros, E. G. (1980) *Phytochemistry* **19**, 2051.
- Schteingart, C. D. and Pomilio, A. B. (1981) *Phytochemistry* **20**, 2589.
- Ariza Espinar, L. (1982) *Bol. Soc. Argent. Bot.* **20**, 267.
- Heywood, V. H., Harborne, J. B. and Turner, B. L. (1977) *The Biology and Chemistry of the Compositae*. Academic Press, London.
- Kasai, R., Okihara, M., Asakawa, J., Mizutani, K. and Tanaka, O. (1979) *Tetrahedron* **35**, 1427.
- Schneider, J. J. and Bhacca, N. S. (1969) *J. Org. Chem.* **34**, 1990.
- Higuchi, R., Kawasaki, T., Biswas, M., Pandey, V. and Dasgupta, B. (1982) *Phytochemistry* **21**, 907.
- Schels, H., Zinmeister, H. D. and Pfeleger, K. (1978) *Phytochemistry* **17**, 523.
- Tsuyuki, T., Hamada, Y., Honda, T., Takahashi, T. and Matsushita, K. (1979) *Bull. Chem. Soc. Japan* **52**, 3127.
- Gorin, P. A. J. and Mazurek, M. (1975) *Can. J. Chem.* **53**, 1212.
- Ritchie, R. G. S., Cyr, N., Korsch, B., Koch, H. J. and Perlin, A. S. (1975) *Can. J. Chem.* **53**, 1424.
- Seo, S., Tomita, Y., Tori, K. and Yoshimura, Y. (1978) *J. Am. Chem. Soc.* **100**, 3331.
- Seo, S., Tomita, Y., Tori, K. and Yoshimura, Y. (1980) *J. Am. Chem. Soc.* **102**, 2512.
- Markham, K. R., Ternai, B., Stanley, R., Geiger, H. and Mabry, T. J. (1978) *Tetrahedron* **34**, 1389.
- Yamasaki, K., Kohda, H., Kobayashi, T., Kasai, R. and Tanaka, O. (1976) *Tetrahedron Letters* 1005.
- Klyne, W. (1950) *Biochem. J.* **47**, xli.
- Hakomori, S. (1964) *J. Biochem. (Tokyo)* **55**, 205.
- Hariharan, V. (1974) *Indian J. Chem.* **12**, 447.
- Ortega, A., Vargas, C., Guerrero, C. and Romo de Vivar, A. (1973) *Rev. Latinoam. Quim.* **4**, 1.
- Bohlmann, F. and Lonitz, M. (1978) *Chem. Ber.* **111**, 843.
- Bohlmann, F. and Lonitz, M. (1980) *Chem. Ber.* **113**, 2410.